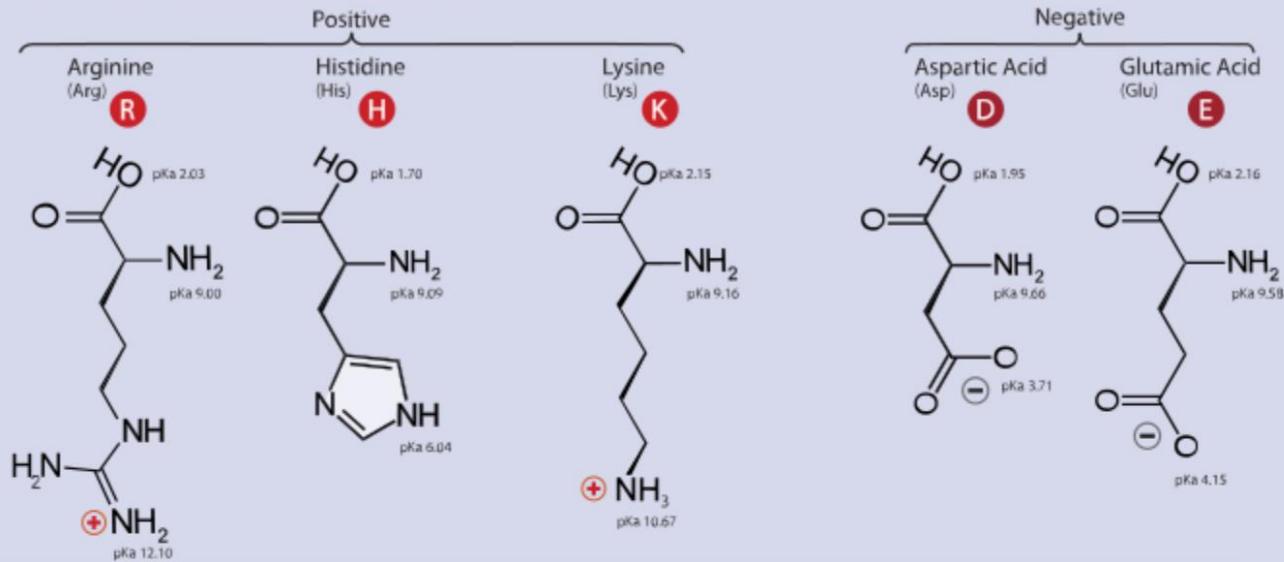


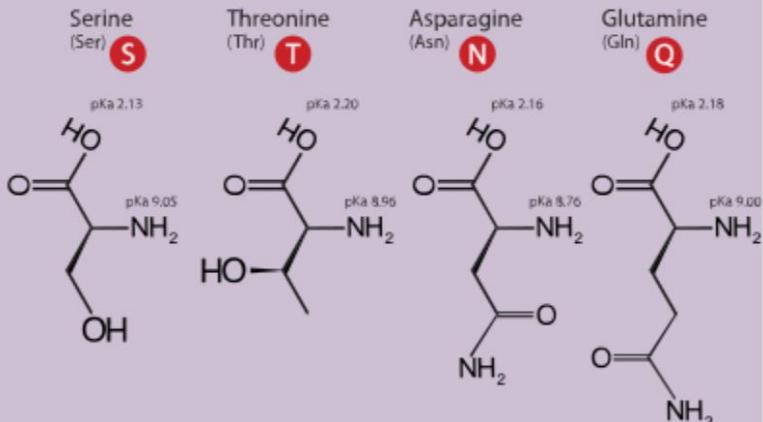
## Twenty-One Amino Acids

⊕ Positive      ⊖ Negative  
 • Side chain charge at physiological pH 7.4

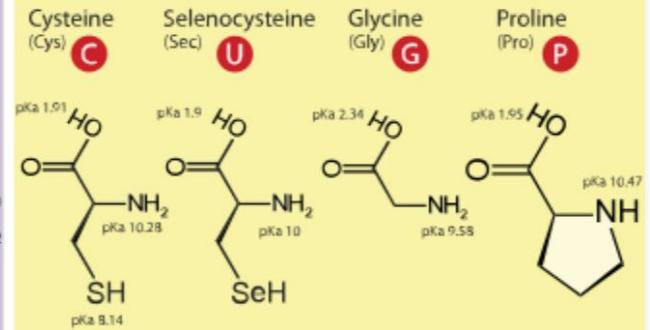
### A. Amino Acids with Electrically Charged Side Chains



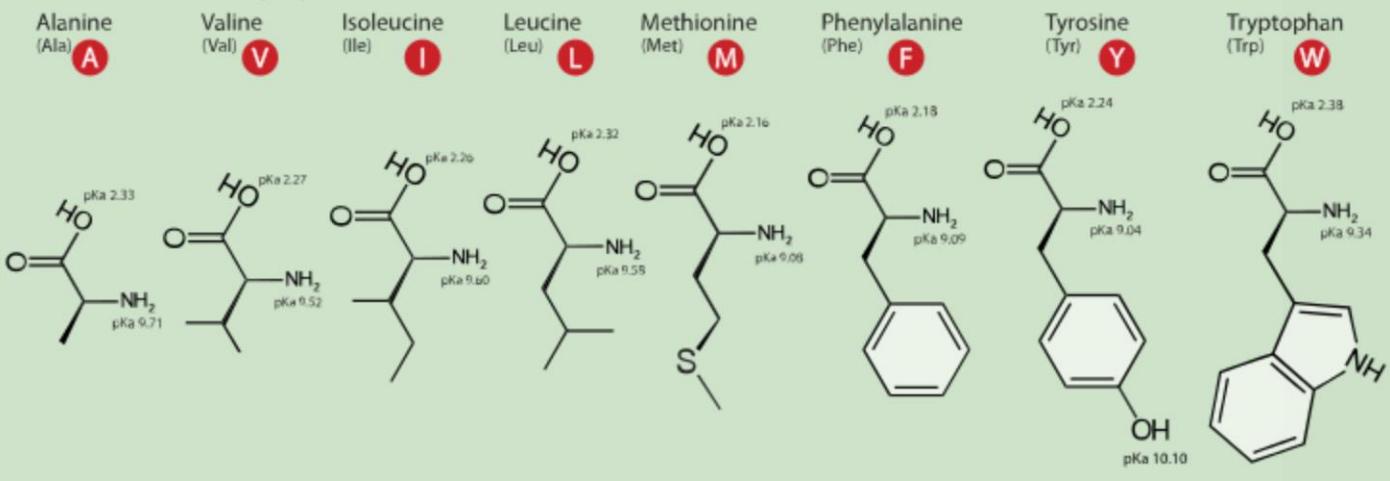
### B. Amino Acids with Polar Uncharged Side Chains

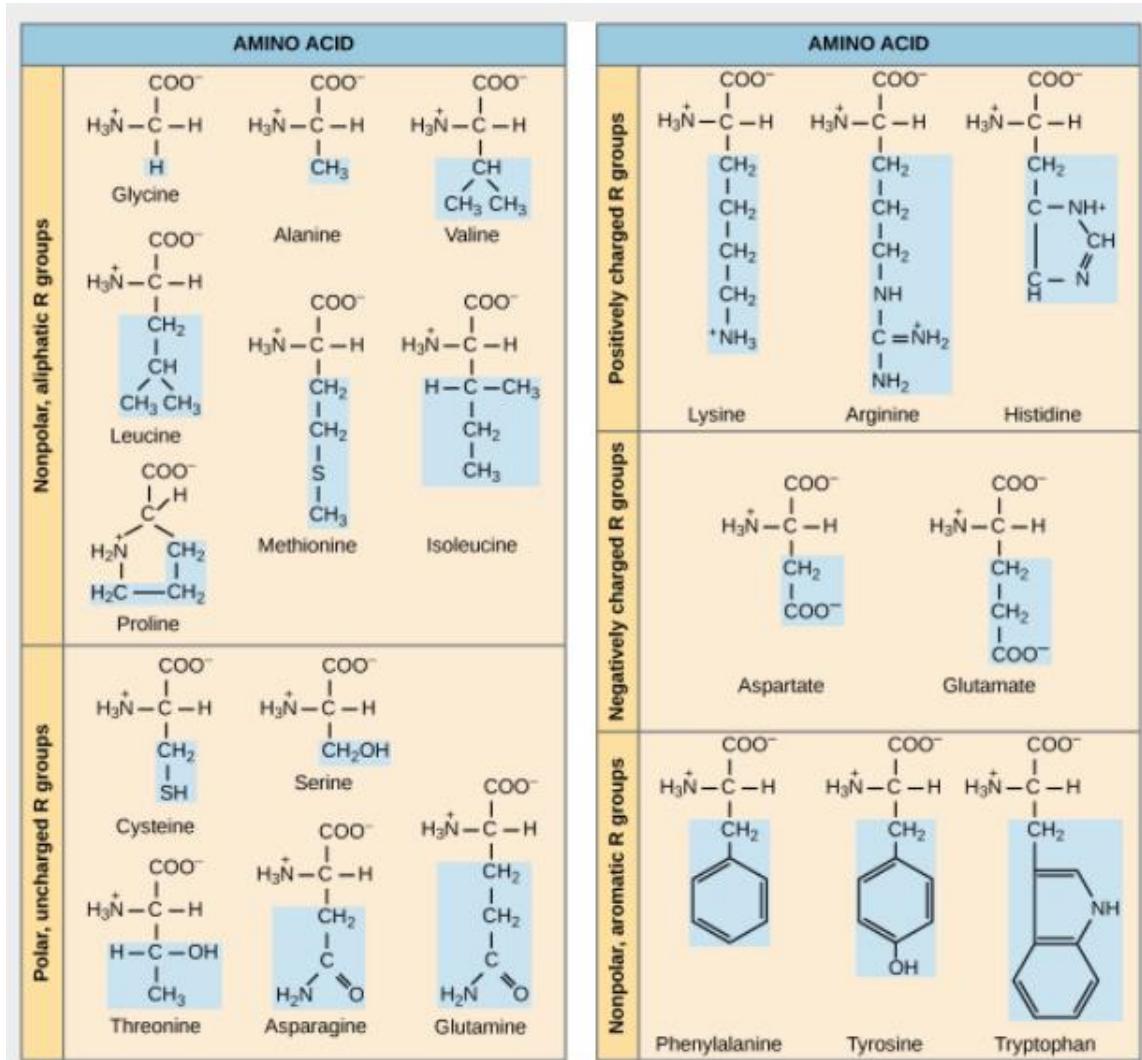


### C. Special Cases



### D. Amino Acids with Hydrophobic Side Chain





Functional Group	Structure	Properties
Hydroxyl	$\text{R} - \text{O} - \text{H}$	Polar
Methyl	$\text{R} - \text{CH}_3$	Nonpolar
Carbonyl	$\text{R} - \text{C}(=\text{O}) - \text{R}'$	Polar
Carboxyl	$\text{R} - \text{C}(=\text{O}) - \text{OH}$	Charged, ionizes to release $\text{H}^+$ . Since carboxyl groups can release $\text{H}^+$ ions into solution, they are considered acidic.
Amino	$\text{R} - \text{N}(\text{H})_2$	Charged, accepts $\text{H}^+$ to form $\text{NH}_3^+$ . Since amino groups can remove $\text{H}^+$ from solution, they are considered basic.
Phosphate	$\text{R} - \text{P}(\text{O})(\text{OH})_2$	Charged, ionizes to release $\text{H}^+$ . Since phosphate groups can release $\text{H}^+$ ions into solution, they are considered acidic.
Sulphydryl	$\text{R} - \text{S}(\text{H})$	Polar

**Figure 2.29** The functional groups shown here are found in many different biological molecules.